



Solubility Behavior of Organic Compounds (Techniques of Chemistry)

David J. W. Grant, Takeru Higuchi

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The role of specific molecular interactions in influencing the solubility behavior of organic compounds are examined, particularly the role of hydrogen bonding. Shows how specific interactions can be used to elicit preferential solubility. Emphasizes interactions occurring in environments of low polarity and explains and predicts solubility phenomena in self-associated solvents. Also considers the kinetics of diffusion and dissolution.

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